



## The influence of natural and anthropogenic secondary sources on the glyoxal global distribution

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### ► To cite this version:

S. Myriokefalitakis, M. Vrekoussis, K. Tsigaridis, F. Wittrock, A. Richter, et al.. The influence of natural and anthropogenic secondary sources on the glyoxal global distribution. *Atmospheric Chemistry and Physics Discussions*, 2008, 8 (1), pp.1673-1708. hal-00328262

**HAL Id: hal-00328262**

**<https://hal.science/hal-00328262>**

Submitted on 10 Oct 2008

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**Secondary sources  
and distribution of  
glyoxal**

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# The influence of natural and anthropogenic secondary sources on the glyoxal global distribution

**S. Myriokefalitakis<sup>1</sup>, M. Vrekoussis<sup>2</sup>, K. Tsigaridis<sup>3</sup>, F. Wittrock<sup>2</sup>, A. Richter<sup>2</sup>,  
C. Brühl<sup>4</sup>, R. Volkamer<sup>5</sup>, J. P. Burrows<sup>2</sup>, and M. Kanakidou<sup>1</sup>**

<sup>1</sup>Environmental Chemical Processes Laboratory, Department of Chemistry, University of Crete, Voutes Campus, P.O. Box 2208, 71003 Heraklion, Greece

<sup>2</sup>Institute of Environmental Physics and Remote Sensing, IUP, University of Bremen, Germany

<sup>3</sup>Laboratoire des Sciences du Climat et de l'Environnement (LSCE), CNRS-CEA, 91191 Gif-sur-Yvette, France

<sup>4</sup>Max Planck Institute, Atmospheric Chemistry Division, Mainz, Germany

<sup>5</sup>Department of Chemistry and Biochemistry, University of Colorado, Boulder, Colorado, USA

Received: 4 December 2007 – Accepted: 16 December 2007 – Published: 31 January 2008

Correspondence to: M. Kanakidou (mariak@chemistry.uoc.gr)

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## Abstract

Glyoxal, the smallest dicarbonyl, which has recently been observed from space, is expected to provide indications on VOC oxidation and secondary aerosol formation in the troposphere. Glyoxal is known to be mostly of natural origin and is produced during biogenic VOC oxidation. However, a number of anthropogenically emitted hydrocarbons, like acetylene and aromatics, have been positively identified as glyoxal precursors. The present study investigates the contribution of pollution emissions to the glyoxal levels by taking into account only the secondary chemical formation of glyoxal from precursors emitted from biogenic, anthropogenic and biomass burning sources. For this purpose, a global 3-dimensional chemistry transport model of the troposphere (TM4) able to simulate the gas phase chemistry coupled with all major aerosol components is used. The model results are compared with satellite observations of glyoxal columns over hot spot areas. According to TM4 model results, the anthropogenic contribution to the glyoxal columns is found to reach 70% in the industrialized areas of the northern hemisphere and up to 20% in the tropics. It is on average three times larger than the secondary production of glyoxal from biomass burning sources. The chemical production of glyoxal is calculated to equal about  $56 \text{ Tg y}^{-1}$  with 70% produced from biogenic hydrocarbons oxidation, 17% from acetylene, 11% from aromatic chemistry, and 2% from ethene and propene. Glyoxal is destroyed by reactions mainly with OH radicals (22%) and by photolysis (65%), but it is also removed from the atmosphere through wet (11%) and dry deposition (6%). Secondary organic aerosol potential formation through glyoxal losses on/in aerosols and clouds was neglected here due to the significant uncertainties associated with the underlying chemistry. The global annual mean glyoxal burden and lifetime in the model domain are estimated at 0.02 Tg and 3 h, respectively.

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## 1 Introduction

In the presence of nitrogen oxides (NO<sub>x</sub>), the photochemical degradation of volatile organic compounds (VOCs) leads to secondary gaseous and particulate products, like ozone (O<sub>3</sub>) and secondary organic aerosols (SOA) that are important contributors to the 'photochemical smog' (Poisson et al., 2000; Kanakidou et al., 2005). Air pollution has severe negative effects on human health (Evans et al., 2002; Pope and Dockery, 2006), ecosystem (Gregg et al., 2003) and regional climate (Ramanathan and Crutzen, 2003). The improvement of air quality in urban areas requires a detailed survey of pollutants like VOCs. Due to the large number of VOC compounds and their complex chemical degradation in the troposphere, significant effort is being put into the assignment of indicators of VOC chemistry that are relatively small molecules, like formaldehyde (HCHO). These gases are affected by both anthropogenic and natural emissions and can be measured in the troposphere.

Recent studies have shown that glyoxal (CHOCHO), the simplest alpha dicarbonyl organic compound, is one of the most prevalent dicarbonyls in the atmosphere. Glyoxal has primary sources from biofuel burning (Hays et al., 2002) and vehicle emissions (Kean et al., 2001; Grosjean et al., 2001) that are highly uncertain due to the large variability of emission factors and the scarcity of the available observations. During burning of biofuels, glyoxal emission rates 3 to 5 times higher than those of formaldehyde have been reported (Hays et al., 2002). During the MCMA-2003 campaign in Mexico City, highly time-resolved direct glyoxal observations showed no accumulation of glyoxal during rush-hour, but a fast rise in glyoxal concentrations after the onset of photochemical activity, indicating that ambient glyoxal is dominated by VOC photochemical sources. It was concluded that glyoxal is a suitable indicator molecule for VOC oxidation processes (Volkamer et al., 2005b).

Glyoxal is produced via the oxidation of numerous hydrocarbons (Calvert et al., 2000, 2002) that are emitted from anthropogenic, biogenic and biomass burning sources. It is a first generation product of acetylene oxidation by OH and is produced from the

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oxidation of a number of other unsaturated VOCs by  $O_3$  and OH, including ethene and isoprene (for a compilation of glyoxal precursor VOCs see Volkamer et al., 2005a and Volkamer et al., 2007). Glyoxal has been identified as first generation product of the aromatic ring-opening routes during the hydroxyl (OH) radical initiated oxidation of aromatic hydrocarbon (Volkamer et al., 2001). The secondary production of glyoxal from the atmospheric degradation of furanones, unsaturated  $\gamma$ -dicarbonyls and phenols has been observed to be negligible (Volkamer et al., 2001; Bloss et al., 2005). The glyoxal formation from isoprene, globally the largest source for secondary glyoxal, mostly proceeds via the further oxidation of the intermediates like glycolaldehyde and hydroxyacetone (Saunders et al., 2003).

Glyoxal has also been detected in atmospheric aerosols (Kawamura et al., 1996; Matsunaga et al., 2003; Liggio and McLaren, 2003; Garcia-Alonso et al., 2006). It was suggested that glyoxal is strongly involved in the formation of SOA in the troposphere by participating in heterogeneous reactions that form organic sulfate compounds (Liggio et al., 2005a). Glyoxal is more reactive with respect to hydration, polymerization and acetal/hemiacetal formation in the presence of alcohols than most other aldehydes (Jang and Kammer, 2001). Based on kinetic studies by Liggio et al. (2005b), the amount of SOA potentially formed by glyoxal could be large enough to account for a significant part of the missing SOA in the low and free troposphere currently not reproduced by models (Heald et al., 2005; Volkamer et al., 2006a). Laboratory measurements consistently find evidence for SOA formation from glyoxal (Liggio et al., 2005a, b; Kroll et al., 2005) but are inconclusive as to the atmospheric relevance as a SOA source. Recent first field evidence of a missing glyoxal sink is consistent with the upper values of SOA formation from glyoxal (Volkamer et al., 2007) and suggests the atmospheric relevance of this process.

Glyoxal is absorbing in the UV and visible spectral range between 200 and 470 nm with two main absorption bands: a broad UV band between 220 and 350 nm and a more structured band between 350 and 470 nm (Atkinson et al., 2006 and references therein). The spectrum between 425 and 457 nm includes a sharp peak of glyoxal at

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455 nm (Volkamer et al., 2005c). This band has enabled the recent direct detection of glyoxal on local (Volkamer et al., 2005b; Sinreich et al., 2007) and global scales (Wittrock et al., 2006), i.e., by the SCIAMACHY satellite instrument, using the well established Differential Optical Absorption Spectroscopy (DOAS) technique.

5 Simultaneous observations of HCHO, CHOCHO, carbon monoxide (CO) and nitrogen dioxide (NO<sub>2</sub>) have been proposed to provide insight to the contribution of photochemistry versus that of primary sources to the building up of pollutants in the troposphere (Garcia-Alonso et al., 2006). In particular, HCHO is used as indicator of VOC oxidation (Palmer et al., 2003; Wittrock et al., 2006). However, over urban hot  
10 spot areas, HCHO is affected by direct emissions and photochemical sources. The ambient levels of CHOCHO are about 10 times less affected by primary sources than HCHO (as argued in Volkamer et al., 2005b). CHOCHO is therefore a better indicator of VOC processes (Volkamer et al., 2005b) and particularly of fast photochemistry of anthropogenic VOCs (Spaulding et al., 2003). Carbon monoxide is a tracer of primary  
15 anthropogenic emissions (Garcia-Alonso et al., 2006). Garcia et al. (2006) used the CHOCHO-CO tracer pair for source apportionment of ambient HCHO, and found that 40–70% (depending on the time of day) of the ambient HCHO was emission-related in Mexico City. Furthermore, the ratio of CHOCHO to NO<sub>2</sub> provides insight to the photochemical age of the studied air mass since it is increasing with aging of air masses and  
20 conversion of NO<sub>2</sub> to nitric acid that is further removed from the atmosphere (Volkamer et al., 2005b; Sinreich et al. 2007).

In the present study, we applied the global 3-dimensional chemistry transport model of the troposphere TM4 in order to investigate glyoxal formation from biogenic, biomass burning and anthropogenic precursors and its seasonal variability. The model results  
25 are constrained by satellite observations from the SCIAMACHY instrument taken in 2005. The potential utility of synergistic use of glyoxal and other short-lived trace gases observations from space to distinguish between the impact of anthropogenic and of natural emissions on tropospheric chemistry is also examined.

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2 Model description

The model used for the present study is the well documented off-line chemical model TM4 (van Noije et al., 2004, 2006). TM4 has been evaluated in the frame of some intercomparison exercises (Shindell et al., 2006; Stevenson et al., 2006). It has 31 vertical hybrid layers from the surface to 10 hPa and can run in two different horizontal resolutions, the low resolution of 6°×4° in longitude and latitude and the high resolution of 3°×2°. For the present study the low resolution version of the model has been mainly used for the source apportionment analysis. The model's input meteorology comes from the ECMWF operational data for 2005 and is updated every 6 h.

2.1 Emissions

The adopted emissions of primary particles (carbonaceous aerosols, dust, sea-salt, small fraction of sulfate) in TM4 are presented in detail by Tsigaridis et al. (2006) and Tsigaridis and Kanakidou (2007). For the biogenic VOC emissions TM4 uses the 1°×1° gridded emission distribution from the POET database (Granier et al., 2005) that is based on the GEIA (Guenther et al., 1995) inventory. For the anthropogenic VOC, NOx and CO emissions, TM4 also uses the POET emission database. This corresponds to the year 2000 that is the most recent year with complete information available. The POET dataset is issued from the Emission Database for Global Atmospheric Research (EDGAR) version 3 inventories for the anthropogenic emissions for the year 1995 with appropriate modifications to refer to recent years (Olivier et al., 2003; Granier et al., 2005). Due to differences in the representation of the volatile organic compound (VOC) chemistry in TM4 compared to the earlier studies by Tsigaridis and coworkers, TM4 uses a different speciation in the VOC emissions. Thus, the present study considers annual biogenic emissions of VOC of 501 Tg of isoprene (1 Tg from the oceans), 244 Tg of monoterpenes and Other Reactive Volatile Organic Compounds (ORVOC) that are able to produce aerosols (like sesquiterpenes, terpenoid alcohols etc, as discussed in Griffin et al., 1999, and incorporated in the model as explained in Kanakidou

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et al., 2000). Additional biogenic emissions of oxygenated organics not considered as aerosol precursors are those of acetone, methanol, acetaldehyde and other light carbonyls ( $27 \text{ Tg yr}^{-1}$ ,  $243 \text{ Tg yr}^{-1}$ ,  $2 \text{ Tg yr}^{-1}$  and  $14 \text{ Tg yr}^{-1}$ , respectively), which are also taken into account in the model. Formaldehyde emissions from biomass burning and anthropogenic activities of  $2.5 \text{ Tg yr}^{-1}$  and  $1.3 \text{ Tg yr}^{-1}$ , respectively, are also considered (POET database). The adopted emissions of aromatic hydrocarbons (POET dataset) total  $33 \text{ Tg yr}^{-1}$  with about 7% emitted from biomass burning, and consist of  $7 \text{ Tg yr}^{-1}$  of benzene,  $14 \text{ Tg yr}^{-1}$  of toluene and  $12 \text{ Tg yr}^{-1}$  of xylene. Among the other compounds of interest for the present study, ethene, propene and acetylene emissions from the same database account for  $17 \text{ Tg yr}^{-1}$ ,  $7 \text{ Tg yr}^{-1}$  and  $7 \text{ Tg yr}^{-1}$ , respectively.

## 2.2 Chemistry

The model considers sulphur, ammonia and  $\text{C}_1\text{-C}_5$  hydrocarbons chemistry including isoprene, as well as highly simplified terpenes and aromatics chemistry. It uses an updated version of the Kanakidou and Crutzen (1999) and Poisson et al. (2000) chemical mechanism suitable to simulate the chemistry of the global troposphere. Gas-phase chemistry and secondary aerosol formation are coupled and computed on-line together with all other major aerosol components (sulphate, nitrate, ammonium, carbonaceous, dust, sea-salt, aerosol water). For the present simulations, dust and sea-salt components have been neglected in order to save computing time. The VOC scheme explicitly considers methane, ethane, ethene, acetylene, propane, propene, butane and isoprene. Reactions of monoterpenes and aromatics are affecting both gas-phase chemistry and aerosol production as described in Tsigaridis and Kanakidou (2003, 2007). Monoterpenes and ORVOC are represented by  $\alpha$ - and  $\beta$ -pinene, while aromatic compounds are represented by benzene, toluene and xylene. The  $\alpha$ - and  $\beta$ -pinene oxidation by hydroxyl (OH) and nitrate ( $\text{NO}_3$ ) radicals and ozone are considered to form semi-volatile organics, which are the precursors of SOA, and  $\text{C}_5$  peroxy radicals, like the first generation peroxy radicals from isoprene oxidation. The reactions of all peroxy radicals with hydrogen peroxide ( $\text{HO}_2$ ) and methyl peroxide ( $\text{CH}_3\text{O}_2$ )

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are taken into account as indicated in Vrekoussis et al. (2004) with updated kinetics based on Atkinson et al. (2006). NO<sub>3</sub> radical reactions with aldehydes, alcohols, dimethylsulfide, butane and unsaturated hydrocarbons are considered. Aromatics are considered to react with OH, O<sub>3</sub> and NO<sub>3</sub>, with the reaction of OH being the dominant removal process accounting for 100, 99.9, 99.8% of the removal of benzene, toluene and xylene, respectively. The reactions of aromatics with OH are considered to produce glyoxal as first generation product and peroxy radicals that can further react to form gaseous and aerosol products. For simplicity purposes their gaseous products are treated in the model as butyloperoxy radicals; attention is paid to preserve the carbon balance. SOA production by VOC oxidation is parameterized as detailed in Tsigaridis and Kanakidou (2003). The full list of chemical reactions and reaction rates adopted in TM4 is given in the Supplementary material (<http://www.atmos-chem-phys-discuss.net/8/1673/2008/acpd-8-1673-2008-supplement.pdf>).

In particular for glyoxal, the TM4 model considers primary formation during the oxidation of isoprene, acetylene, aromatic hydrocarbons (Volkamer et al., 2005a) and ozonolysis of ethene and propene as well as secondary formation (via glycolaldehyde – HOCH<sub>2</sub>CHO) from the oxidation of ethene and from peroxy radicals produced during the oxidation of isoprene (Table 1). Other biogenic hydrocarbons, like terpenes, and other reactive organics, like sesquiterpenes and terpenoid alcohols, also produce glyoxal (as secondary product) in TM4, since their chemistry is lumped on the first generation peroxy radicals of isoprene (as parameterized in Tsigaridis and Kanakidou, 2003). Glyoxal is destroyed by photolysis (Atkinson et al., 2006; Volkamer et al., 2005c; Tadic et al., 2006), reaction with OH (Feierabend et al., 2008) and to a lesser extent with NO<sub>3</sub>, and via wet and dry deposition. For the dry deposition of glyoxal as for the other gaseous species, the Ganzeveld and Lelieveld (1995) resistance analogy scheme has been used, which is based on the formulation developed by Wesely (1989). For the wet deposition, both large scale and convective precipitation are considered and the in-cloud and below cloud scavenging are parameterised in the model TM4 as described in detail by Jeuken et al. (2001). In-cloud scavenging of glyoxal is

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parameterized based on the Henry law coefficient of  $K_H=2.6\times 10^7\text{ M/atm}$  (Kroll et al., 2005) that is 2 orders of magnitude higher than the Henry Law coefficient suggested by Lim et al. (2005). The in-cloud loss of glyoxal has been suggested to be a significant sink for glyoxal leading to SOA formation (Liggio et al., 2005; Loeffler et al., 2006; Sorooshian et al., 2007). The importance of this process for SOA budget in the global troposphere is under investigation by additional 3-dimensional simulations and will be the subject of a forthcoming publication.

### 2.3 The simulations

In order to investigate the influence of natural and anthropogenic emissions of hydrocarbons on glyoxal distribution and its global budget, three different glyoxal-chemistry scenarios have been considered. Any potential primary emissions of glyoxal are neglected in the present study due to the large variability and uncertainty in glyoxal emission factors. The first simulation (S1) is considering glyoxal production only from the oxidation of the naturally emitted VOCs. S1 also considers in-cloud loss of glyoxal as explained in Sect. 2.2. The second simulation (S2) is as S1 but also takes into account the contribution of biomass burning emissions of ethene, propene, acetylene and aromatics to the chemical production of glyoxal. The third simulation (S3) is as (S2) but considers in addition the contribution of anthropogenic emissions of ethene, propene, acetylene and aromatics to glyoxal formation, i.e. accounts for all known photochemical sources of glyoxal. This simulation has been also performed at the high horizontal resolution of TM4 (S3-H). The difference between the simulations (S2) and (S1) is attributed to the secondary production of glyoxal from the biomass burning VOCs; whereas that between (S3) and (S2) is attributed to the production of glyoxal from anthropogenic VOC. Primary emissions of glyoxal have been neglected in the present study due to the large uncertainty in their intensity resulting from the significant variability in the emission factors and the scarcity of available data.

When compared with the earlier work by Wittrock et al. (2006), the present study

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i) uses updated anthropogenic and biomass burning emission inventories based on the POET database for the year 2000 and about 20% higher isoprene emissions of  $500 \text{ Tgyr}^{-1}$  that is within the range of the present-day estimates by Guenther et al. (2006), ii) takes into account formation of glyoxal also from ethene, propene, acetylene and aromatics that are predominantly of anthropogenic origin but have also a minor natural component, iii) considers the temperature dependence of the reaction of glyoxal with OH radicals (Feierabend et al., 2008), iv) calculates on line as described by Landgraf and Crutzen (1998) and Lelieveld et al. (2002) the photolysis rates of glyoxal based on the IUPAC and JPL recommendations (Atkinson et al., 2006 and references therein; Sander et al., 2006). Moreover, Wittrock et al. (2006) used diurnal mean glyoxal columns to compare with the satellite observations whereas in the present study the columns calculated for the satellite overpass time have been used (10:00–11:00 a.m. local time).

### 3 Satellite observations

#### 3.1 The instrument

The sensor SCIAMACHY (SCanning Imaging Absorption spectroMeter for Atmospheric CartographY) is mounted on the ESA ENVISAT satellite. It is designed to measure sunlight, transmitted, reflected and scattered by the earth's atmosphere or surface in the ultraviolet, visible and near infrared (Burrows et al., 1995; Bovensmann et al., 1999). The sun-synchronous, near polar orbit of the satellite has a local equator crossing time of 10:00 a.m. in the descending node. The alternative measurements of nadir and limb viewing geometries (Gottwald et al., 2006) are coupled with a swath width of 960 km and global coverage is achieved within six days. The ground pixel size used in this study is 30 km along track by 60 km across track.

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3.2 The retrieval technique and glyoxal algorithm

The retrieval approach adopted for the vertical columns of CHOCHO is based on the well established DOAS technique. Briefly, DOAS is used for the determination of the Slant Column densities (SCD) of the respective absorbers in a given spectral window (436–457 nm for the present study). Only ground scenes having less than 20 percent cloud cover are considered. The SCD is then converted to Vertical Column Densities (VCD) by applying the air mass factor (AMF), which is defined as the ratio of the SCD/VCD. This accounts for the path of light through the atmosphere and considers the vertical profiles of scattering and absorbing species (Wittrock et al., 2004). The absorption cross sections of CHOCHO (Volkamer et al., 2005c), O<sub>3</sub>, NO<sub>2</sub>, H<sub>2</sub>O, O<sub>4</sub>, chlorophyll (Bracher and Tilzer, 2001; Vountas et al., 2007), a ring spectrum, which accounts for both rotational and vibrational Raman scattering, and a quadratic polynomial are included in the fitting procedure.

Prior conversion to vertical columns, the retrieved slant columns were normalised by assuming a mean value of  $0.5 \times 10^{13}$  molecules cm<sup>-2</sup> for CHOCHO over the Pacific in latitudes (30° N–30° S) in the region between 165° E and 165° W. This normalisation is necessary for many GOME and SCIAMACHY products to compensate for offsets introduced by the solar reference measurements and interference from other absorbers (Richter and Burrows, 2002).

4 Results and discussion

4.1 Comparison against glyoxal surface observations

Even at its high spatial resolution of 2° latitude and 3° longitude, TM4 has a rather coarse grid that at around 45° N corresponds to about 220 km in latitude × 230 km in longitude. Therefore, it is expected that as most global models, TM4 underestimates short lived species concentrations over hot spot areas. This is because these hot spot

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areas extend only over a small fraction of the surface of the model grid boxes and thus in these model grids the short lived species are artificially diluted over hot spot areas by instantaneous mixing with cleaner air from the surrounding locations in the same grid boxes. Keeping in mind these limitations, the results of the high resolution TM4 simulation (S3-H) are evaluated against ground based observations reported in the literature for different places of the world (Table 2). An overall reasonable agreement is seen between the TM4 S3-H simulated boundary layer glyoxal concentrations and the surface observations. The model reproduces remarkably well the observations by Sinreich et al. (2007) at coastal oceanic areas on board the RV Ron Brown. In contrast, it misses the extremely high values observed inside the urban core of Hong Kong by Ho and Yu (2002).

In addition to the above mentioned spatial effects that contribute to the differences between model results and observations, uncertainties in oxidant fields, VOC loadings and particulate matter levels are also at the origin of differences illustrated in Table 2. In addition, potentially existent primary glyoxal sources that have been neglected in TM4 (as discussed in Sect. 2.3) could be a reason for an underestimate of glyoxal levels by the model. On the opposite, a potentially strong missing sink of glyoxal could explain an overestimate of glyoxal in the model. Such sink of glyoxal on aerosols has been suggested by Volkamer et al. (2007a) to reconcile glyoxal observations in Mexico City with results of a box model that was very well experimentally constrained.

It has to be stressed that there is a clear need for more ground based glyoxal observations available for model and satellite evaluations, in particular over oceanic areas since most of the glyoxal observations are restricted to continental anthropogenically influenced locations.

## 4.2 Annual mean glyoxal columns

Focusing on the global annual mean CHOCHO column distribution for the year 2005, the S3-H simulations are compared with the columns retrieved from SCIAMACHY measurements in Fig. 1a. The model captures high columns of CHOCHO in the tropics as

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sociated mainly with biogenic and biomass burning emissions as has been observed by SCIAMACHY (Fig. 1b), although the model underestimates the observations.

As seen in the figures, TM4 misses the significant CHOCHO levels observed by SCIAMACHY over the tropical oceans and close to upwelling areas. This statement is also supported by the correlations shown in Fig. 1c. In this figure the simulated annual mean glyoxal vertical columns have been plotted against the SCIAMACHY annual mean observations in 2005. The points correspond to all  $2^\circ \times 3^\circ$  grid boxes with SCIAMACHY observations. To better visualize the highly scattered data the modeled CHOCHO columns were grouped per  $2.5 \times 10^{13}$  molecules  $\text{cm}^{-2}$  of SCIAMACHY values. Thus, the binned data shown as solid blue squares have been derived and the blue line corresponds to the linear fit of these data that indicates a slope of 0.43 and a correlation coefficient  $r$  of 0.95. When neglecting the observations and the simulated values over the oceans the solid black circles have been derived that correspond to better correlation ( $r=0.99$ ) and slope of 0.62 that is closer to unity than when considering the oceanic values.

This may indicate the existence of primary or secondary tropical sources of glyoxal over the oceans that are neglected or underestimated by the model, although some outflow from the continents (Sinreich et al., 2007) and slight enhancement over the tropical oceans and the north Atlantic is currently simulated by TM4 compared to the earlier simulations by Wittrock et al. (2006). These patterns will be investigated further in a forthcoming paper.

Focusing on the model results over the continents, in Fig. 1d the binned data derived as above explained are plotted against the SCIAMACHY annual mean observations for both the high (S3-H) and the low (S3) resolution simulations considering all photochemical sources of glyoxal as well as for the S2 simulation that neglects the anthropogenic contribution to this source. From this comparison it can be seen that the S3-H high resolution simulation performs the best (slope=0.62,  $r=0.99$ ) when compared to the lower resolution S3 simulations (slope=0.60,  $r=0.95$ ) and the S2 simulation that neglects the anthropogenic contribution (slope=0.53,  $r=0.95$ ). These comparisons suggest i) that

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TM4 underestimates the annual mean glyoxal columns observed by SCIAMACHY in 2005, ii) that accounting for the anthropogenic contribution to the photochemical formation of glyoxal, TM4 results compare better with the observations.

Figure 2a depicts the global annual mean CHOCHO column distribution for the year 2005, computed by the S3 simulation to be compared with Fig. 2c that shows the SCIAMACHY observations at the same spatial resolution ( $4^{\circ} \times 6^{\circ}$ ). Figure 2b depicts the calculated annual mean column of glyoxal produced from biogenic VOC oxidation (simulation S1). Most of the annual mean column of glyoxal over the tropics that maximizes at about  $5 \times 10^{14}$  molecules  $\text{cm}^{-2}$  is attributed to the oxidation of biogenic VOC. Annual mean columns smaller than  $3 \times 10^{14}$  molecules  $\text{cm}^{-2}$  are computed over northern hemisphere mid-latitude continental regions. Note that these biogenically derived glyoxal columns are slightly smaller than the earlier simulations in Wittrock et al. (2006) that did not account for the wet removal of glyoxal in-clouds and used slower photolysis rate of glyoxal than the present study.

As mentioned in Sect. 2.3, biomass burning primary emissions of glyoxal in the tropics (and elsewhere) have been neglected in the present study since they are highly uncertain, although they might be of importance for glyoxal columns in this region. However, the secondary biomass burning source of glyoxal from the oxidation of VOC is taken into account in the simulations (S2 and S3). Its contribution to the glyoxal column is quantified as the difference between the simulations S2 and S1 and is depicted in Fig. 2e (annual mean column) and Fig. 3i, j (seasonally mean columns). According to our calculations the secondary source of glyoxal from biomass burning contributes about  $3 \text{ Tg yr}^{-1}$  to the global chemical production of glyoxal and therefore increases its global annual mean burden by about 5%, adding at most  $2 \times 10^{14}$  molecules  $\text{cm}^{-2}$  to the glyoxal column over tropical biomass burning areas.

Simulation S3 calculates significant glyoxal columns of about 2 to  $4 \times 10^{14}$  molecules  $\text{cm}^{-2}$  over the anthropogenic hot spot areas of USA, Europe, India and China (Fig. 2a) that are also detectable from space (Fig. 2c) but are not reproduced in simulations S1 and S2. Figure 2d depicts the increase in the annual

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mean column of glyoxal due to the consideration of glyoxal production during the oxidation of anthropogenically emitted VOC (difference between simulations S3 and S2). According to our calculations, the secondary source of glyoxal from anthropogenic emissions contributes about  $10 \text{ Tgyr}^{-1}$  to the global chemical production of glyoxal, i.e., that is about 18% of the overall chemical production. It is interesting to note that the glyoxal global annual mean burden issued from the anthropogenic emissions is about 32% of the total; this fraction is larger than the corresponding fraction of the chemical production. This reflects the fact that the anthropogenic source is located mostly in the extratropics where the destruction of glyoxal by photolysis and reaction with OH radical is moderate and thus glyoxal lifetime is longer than its global mean value.

On the contrary, the biogenically derived burden of glyoxal is 63% of the total glyoxal burden whereas the corresponding chemical production is higher, about 77% of the total. This increased chemical production is compensated by a shorter lifetime of glyoxal in the tropics. Thus, the computed global lifetime of glyoxal averaged year-around increases from 2.4 h when only biogenic sources are taken into account (S1) and 2.5 h when biomass burning sources are also considered (S2) to 3 h when all sources are taken into account (S3).

Our simulations indicate that the global mean lifetime of glyoxal critically depends on the location of its sources. In particular, the anthropogenic sources of glyoxal are more effective in increasing its burden since they are located mostly in the mid latitudes where the photochemical sink of glyoxal is moderate and thus its lifetime longer than in the tropics.

### 4.3 Seasonal variation

It is worth comparing the seasonal pattern of the calculated and observed glyoxal columns. Figure 3 depicts the NH winter and NH summer column distributions of glyoxal as observed by SCIAMACHY (panels 3a and b, respectively) and as calculated by TM4 S3 simulation (panels 3c and d, respectively). In the NH, TM4 simulations

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show higher glyoxal columns during summer than during winter in agreement with the satellite observations, reflecting the photochemical origin of glyoxal and the important contribution of the biogenic VOC oxidation to glyoxal chemical production (Fig. 3e and f). It is also remarkable that during NH summer the model overestimates the CHOCHO over the Amazon, but not over the African tropical forests. Biomass burning emissions occurring mostly in the tropics present also a strong seasonality with significantly different geographical patterns from one season to another that are reflected in glyoxal distribution as depicted in Fig. 3i and j. However, their contribution to the glyoxal columns is minor compared to that from the biogenic VOC oxidation (Fig. 3e and f). Therefore, most probably the model overestimate of glyoxal columns is related to the biogenic VOC emission inventory adopted in our model.

Biogenic VOC emissions are enhanced under warm and sunny conditions (Guenther et al., 2006). In particular, biogenic VOC emissions between 30 N and 60 N are about 50 times higher during summer than during winter. On the contrary, the anthropogenic VOC emissions present a much weaker seasonality. In our model simulations this seasonality has been neglected, since the EDGAR and POET emissions database do not provide information on the seasonal variability of the anthropogenic VOC emissions. The calculated hydroxyl radical concentration, the main tropospheric oxidant responsible for glyoxal formation, shows a strong seasonality, with a summer-to-winter ratio that varies near the surface in the northern troposphere mid-latitudes from about 2 to more than 20, as presented in Fig. 4. The strong seasonality in oxidant levels and biogenic emissions leads to about 25% higher global production of glyoxal during the NH summer than during the NH winter (Fig. 3g and h). In parallel, the losses of glyoxal are also about 25% higher during summer as a result of increased losses of glyoxal by photodissociation (20%), by OH radical (40%) and by deposition (25%). The overall result is an almost 10% enhancement of the glyoxal column during NH summer (global burden 0.019 Tg) compared to NH winter (global burden 0.017 Tg). Note that in the present study, a potential aerosol sink of glyoxal that could be of relevance for atmospheric conditions is neglected.

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#### 4.4 Ratios of trace gases columns

Due to the strong anthropogenic signal in the glyoxal column, it is expected that synergistic use of glyoxal and formaldehyde observations from space could provide a powerful tool to evaluate VOC emissions and impact on tropospheric chemistry as a function of source region and type. This is further investigated by examining the computed ratios of CHOCHO to HCHO over the tropics and over the hot-spot northern hemisphere areas. The satellite observations of CHOCHO and HCHO molecules were used to compute the respective ratio above the 30 most populated areas of the world (<http://www.infoplease.com/ipa/A0884418.html>). The mean value of the ratio of CHOCHO/HCHO above these regions was  $0.037 \pm 0.014$ . This mean ratio is in excellent agreement with the value of 0.025 observed in Mexico City based on the data provided by Garcia et al. (2006), and taking into account the diurnal profile of these carbonyls to match the satellite overpass time. The same model analysis performed aloft biogenic sources gave a slightly enhanced average value equal to  $0.050 \pm 0.025$ . This latter number is in good agreement with the range of values 0.04–0.06 reported by Munger et al. (1995) and Spaulding et al. (2003). These values are contrasted with those computed by the TM4 model (Fig. 5a): the model calculated ratios, derived from the 10–11 am local time simulations, are in reasonable agreement with the satellite data (not shown) over the selected regions with intense anthropogenic emissions, namely Europe, China, India and the United States. Figure 5b illustrates the same ratio when the anthropogenically produced glyoxal is not taken into account (simulation S2). As expected, the major impact of this photochemical source of glyoxal is seen above the northern hemisphere areas.

The differences in the CHOCHO/HCHO ratio between simulations S3 and S2 that is attributed to the photochemical source of glyoxal from human activities is shown in Fig. 5c in which the anthropogenically influenced regions are clearly identified.

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#### 4.5 Glyoxal annual global budget

According to the TM4 simulations, CHOCHO global annual chemical production totals 56 Tg CHOCHO  $\text{y}^{-1}$ . This results by 70% from the oxidation of biogenics, isoprene and terpenes, among which the predominant portion is formed via formation of glycolaldehyde and hydroxyacetone and subsequent degradation. To a smaller extent CHOCHO is formed from AVOCs, i.e. 17% from acetylene, 11% from aromatics, and 2% from ethene and propene. According to the TM4 model simulations, the contribution of anthropogenic VOC oxidation to CHOCHO column is limited to 3–20% over the tropics and maximizes over the urban-influenced areas in the Northern Hemisphere where it reaches 20–70%; however, the glyoxal column is relatively low ( $\sim 2$  to  $\sim 4 \times 10^{14}$  molecules  $\text{cm}^{-2}$ ) in the anthropogenic influenced regions. This is consistent with findings in Mexico City, where aromatics make some 70–80% of the glyoxal production rate. A direct comparison of the relative contribution of individual VOC sources and glyoxal sinks is possible for Mexico City (data from Table 1 of Volkamer et al., 2007a) and the Pearl River Delta (Volkamer et al., 2007a), in good agreement with the model results in this study.

Glyoxal is destroyed by photolysis (60%), and OH radical reactions (23%), but it is also removed from the atmosphere through wet deposition (11%) and dry deposition (6%). As a result, the glyoxal global burden equals 0.02 Tg and its global mean lifetime 3 h.

It is also remarkable that on an annual basis according to our model simulations, about 5 Tg of glyoxal are removed from the atmosphere via wet deposition processes. This amount has the potential to form secondary organic aerosol as the result of cloud processing of glyoxal, and has to be considered an upper limit for glyoxal contribution to SOA formation from cloud processing in the troposphere. Glyoxal related SOA formation in the absence of clouds as recently indicated may however add further SOA (Volkamer et al., 2007a). This topic requires further investigation and refinement.

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The global 3-dimensional chemistry/transport model TM4 has been applied to simulate the temporal and spatial distribution of CHOCHO column in the global troposphere with focus on the anthropogenic contribution to the CHOCHO levels and seasonality.

5 Aromatic and other anthropogenic hydrocarbons can play an important role for glyoxal columns. Wittrock et al. (2006) showed significant amounts of glyoxal above highly populated regions using measurements by the SCIAMACHY sensor. The enhanced CHOCHO columns above regions with high anthropogenic emissions are confirmed by the latest SCIAMACHY measurements, and ground-based satellite validation efforts  
10 in Mexico City, and the Pearl River Delta, China (Volkamer et al., 2006b). The global annual chemical production of CHOCHO has been calculated by the TM4 model to equal 56 Tg CHOCHO  $\text{y}^{-1}$ . Anthropogenic compounds account for about 30% of the total chemical production with acetylene producing about 17% and aromatic hydrocarbons (toluene, xylene and benzene) about 11%. According to TM4 model calculations,  
15 the contribution of anthropogenic VOC to the CHOCHO column is limited to 3–20% over the tropics and maximizes over urban-influenced areas in the Northern Hemisphere where it reaches 70%. The improvement in the computed glyoxal total column compared to the earlier study by Wittrock et al. (2006) resulted in better agreement between the model results and the glyoxal observations in the mid latitudes. However,  
20 the model does not capture the enhancement of CHOCHO columns over the tropical ocean seen by SCIAMACHY. This may indicate the existence of primary or secondary tropical and/or oceanic sources of CHOCHO that are neglected or underestimated by the model. Our model simulations yield 5 Tg  $\text{yr}^{-1}$  as the upper limit of glyoxal potential contribution to the SOA formation via multiphase chemistry in clouds. This first  
25 estimate deserves careful and further refinement. The role of oceans, clouds and possibly aerosols are areas of major uncertainties for glyoxal modeling and deserve further studies.

*Acknowledgements.* The authors acknowledge support by the ACCENT EU Network of ex-

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cellence. S. Myriokefalitakis acknowledges support by a PENED 03ED373 grant co-financed by EU-European Social Fund (75%) and the Greek Ministry of Development – GSRT (25%). M. Vrekoussis acknowledges consecutive supporting fellowships by the A. v. Humbolt Foundation and the EU (Marie Curie – Intra European Fellowships). We thank K. Feierabend and J. Burkholder for communication of their results prior publication.

## References

- Atkinson, R., Baulch, D. L., Cox, R., A., Crowley, J., N., Hampson, R., F., Hynes, R., G., Jenkin, M., E., Rossi, M., J., Troe, J., and IUPAC Subcommittee: Evaluated kinetic and photochemical data for atmospheric chemistry: Volume II – gas phase reactions of organic species, *Atmos. Chem. Phys.*, 6, 3625–4055, 2006, <http://www.atmos-chem-phys.net/6/3625/2006/>.
- Bloss, C., Wagner, V., Jenkin, M. E., Volkamer, R., Bloss, W. J., Lee, J. D., Heard, D. E., Wirtz, K., Martin-Reviejo, M., Rea, G., Wenger, J. C., and Pilling, M. J.: Development of a detailed chemical mechanism (MCMv3.1) for the atmospheric oxidation of aromatic hydrocarbons, *Atmos. Chem. Phys.*, 5, 641–664, 2005, <http://www.atmos-chem-phys.net/5/641/2005/>.
- Bovensmann, H., Burrows, J. P., Buchwitz, M., Frerick, J., Noel, S., Rozanov, V. V. Chance, K. V., and Goede, A. P. H.: SCIAMACHY: Mission objectives and measurement modes, *J. Atmos. Sci.*, 56, 127–150, 1999.
- Borrego, C., Gomes, P., Barros, N., and Miranda, A. I.: Importance of handling organic atmospheric pollutants for assessing air quality, *J. Chromatogr. A*, 889, 271–279, 2000.
- Bracher, A., and Tilzer, M.: Underwater light field and phytoplankton absorbance in different surface water masses of the Atlantic Sector of the Southern Ocean, *Pol. Biol.*, 24, 687–696, 2001.
- Burrows, J. P., Hölzle, E., Goede, A. P. H., Visser, H., and Fricke, W.: SCIAMACHY – Scanning Imaging Absorption Spectrometer for Atmospheric Chartography, *Acta Astronaut.*, 35, 5 445–451, 1995.
- Calvert, J. G., Atkinson, R., Kerr, J. A., Madronich, S., Moortgat, G. K., Wallington, T. J., and Yarwood, G.: *The Mechanisms of Atmospheric Oxidation of the Alkenes*, Oxford University Press, New York, 2000.

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Calvert, J. G., Atkinson, R., Becker, K. H., Kamens, R. M., Seinfeld, J. H., Wallington, T. J., and Yarwood, G.: The Mechanisms of Atmospheric Oxidation of Aromatic Hydrocarbons, Oxford University Press, Oxford, 2002.

Evans, J., Levy, J., Hammitt, J., Santos-Burgoa, C., Castille-Jos, M., Caballero-Ramirez, M., Hernandez-Avila, M., Riojas-Rodriguez, H., Rojas-Bracho, L., Serrano-Trespalacios, P., Spengler, J. D., and Suh, H.: Health benefits of air pollution control, in: Air Quality in the Mexico Megacities: An Integrated Assessment, edited by: Molina, L. T. and Molina, M. J., Kluwer Academic Publishers, 103–136, 2002.

Feierabend, K., Zhu, L., Talukdar, R. K., and Burkholder, J. B.: Rate coefficients for the OH + HC(O)C(O)H (Glyoxal) reaction between 210 and 390 K, J. Phys. Chem. A, 112(1), 73–82, doi:10.1021/jp0768571, 2008.

Ganzeveld, L. and Lelieveld, J.: Dry deposition parameterization in a chemistry general circulation model and its influence on the distribution of reactive trace gases, J. Geophys. Res., 100, 20 999–21 012, 1995.

García-Alonso, S., Pérez-Pastor, R., and Sevillano-Castaño, M. L.: Determination of glyoxal and methylglyoxal in atmospheric particulate matter by 2,4-dinitrophenylhydrazine derivatization, Toxicol. Environ. Chem., 88(3), 445–452, doi:10.1080/02772240600796837, 2006.

Garcia, A. R., Volkamer, R., Molina, L. T., Molina, M. J., Samuelsson, J., Mellqvist, J., Galle, B., Herndon, S., and Kolb, C. E.: Separation of emitted and photochemical formaldehyde in Mexico City using a statistical analysis and a new pair of gas-phase tracers, Atmos. Chem. Phys., 6, 4545–4557, 2006, <http://www.atmos-chem-phys.net/6/4545/2006/>.

Gottwald, M., Bovensmann, H., Lichtenberg, G., Noel, S., von Bargaen, A., Slijkhuis, S., Piters, A., Hoogeveen, R., von Savigny, C., Buchwitz, M., Kokhanovsky, A., Richter, A., Rozanov, A., Holzer-Popp, T., Bramstedt, K., Lambert, J.-C., Skupin, J., Wittrock, F., Schrijver, H., and Burrows, J.P.: SCIAMACHY, Monitoring the Changing Earth's Atmosphere, Published by DLR, 2006.

Gregg, J. W., Jones, C. G., and Dawson, T. E.: Urbanization, air pollution and tree growth in the vicinity of New York City, Nature, 424, 183–187, 2003.

Granier C., Guenther, A., Lamarque, J.F., Mieville, A., Muller, J. F., Olivier, J., Orlando, J., Peters, J., Petron, G., Tyndall, G., and Wallens, S.: POET, a database of surface emissions of ozone precursors, available on the internet at: <http://www.aero.jussieu.fr/project/ACCENT/POET.php>, 2005.

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- Griffin, R. J., Cocker III, D. R., Flagan, R. C., and Seinfeld, J. H.: Organic aerosol formation from oxidation of biogenic hydrocarbons, *J. Geophys. Res.*, 104, 3555–3567, 1999a
- Griffin, R. J., Cocker III, D. R., Seinfeld, J. H., and Dabdub, D.: Estimate of global atmospheric organic aerosol from oxidation of biogenic hydrocarbons, *Geophys. Res. Lett.* 26, 2721–2724, 1999b
- Grosjean, E., DeAndrade, J. B., and Grosjean, D.: Carbonyl products of the gas-phase reaction of ozone with simple alkenes, *Environ. Sci. Technol.*, 30, 975–983, 1996a.
- Grosjean, E. and Grosjean, D.: The Reaction of Unsaturated Aliphatic Oxygenates with Ozone, *J. Atmos. Chem.*, 32, 205–232, 1999.
- Grosjean, D., Grosjean, E., and Gertler, A. W.: On-Road Emissions of Carbonyls from Light-Duty and Heavy-Duty Vehicles, *Environ. Sci. Technol.*, 35, 45–53, 2001.
- Guenther, A., Hewitt, C. N., Erickson, D., Fall, R., Geron, C., Graedel, T., Harley, P., Klinger, L., Lerdau, M., McKay, W. A., Pierce, T., Scholes, B., Steinbrecher, R., Tallamraju, R., Taylor, J., and Zimmerman, P.: A global model of natural volatile organic compound emissions, *J. Geophys. Res.*, 100, 8873–8892, 1995.
- Guenther, A., Karl, T., Harley, P., Wiedinmyer, C., Palmer, P. I., and Geron, C.: Estimates of global terrestrial isoprene emissions using MEGAN (Model of Emissions of Gases and Aerosols from Nature), *Atmos. Chem. Phys.*, 6, 3181–3210, 2006, <http://www.atmos-chem-phys.net/6/3181/2006/>.
- Hays, D. M., Geron, D. C., Linna, J. K., Smith, D. M., and Schauer, J. J.: Speciation of Gas-Phase and Fine particle emissions from burning of foliar fuels, *Environ. Sci. Technol.*, 36(11), 2281–2295, doi:10.1021/es0111683, 2002.
- Heald, C. L., Jacob, D. J., Park, R. J., Russell, L. M., Huebert, B. J., Seinfeld, J. H., Liao, H., and Weber, R. J.: A large organic aerosol source in the free troposphere missing from current models, *J. Geophys. Res.*, 32, L18809, doi:10.1029/2005GL023831, 2005.
- Ho, S. S. H. and Yu, J. Z.: Feasibility of collection and analysis of airborne carbonyls by on-sorbent derivatization and thermal desorption, *Anal. Chem.*, 74 1232–1240, 2002.
- IUPAC Subcommittee on Gas Kinetic Data Evaluation for Atmospheric Chemistry, Summary of Evaluated Kinetic and Photochemical Data for Atmospheric Chemistry, Web Version February, [http://www.iupac-kinetic.ch.cam.ac.uk/summary/IUPACsumm\\_web\\_Feb2006.pdf](http://www.iupac-kinetic.ch.cam.ac.uk/summary/IUPACsumm_web_Feb2006.pdf), 2006.
- Jeuken, A., Veeffkind, J. P., Dentener, F., Metzger, S., and Gonzalez, C. R.: Simulation of the aerosol optical depth over Europe for August 1997 and a comparison with observations, *J. Geophys. Res.*, 106, 28 295–28 311, 2001.

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- Jang, M. and Kamens, R. M.: Atmospheric secondary aerosol formation by heterogeneous reactions of aldehydes in the presence of a sulfuric acid aerosol catalyst, *Environ. Sci. Technol.*, 35, 24, 4758–4766, 2001.
- Jing, L. H., Steinberg, S. M., and Johnson, B. J.: Aldehyde and monocyclic aromatic hydrocarbon mixing ratios at an urban site in Las Vegas, Nevada, *J. Air Waste Manage.*, 51 1359–1366, 2001.
- Kanakidou, M. and Crutzen, P. J.: The photochemical source of carbon monoxide: Importance, Uncertainties and feedbacks, *Chemosphere: Global Change Science*, 1, 91–109, 1999.
- Kanakidou, M., Seinfeld, J. H., Pandis, S. N., Barnes, I., Dentener, F. J., Facchini, M. C., van Dingenen, R., Ervens, B., Nenes, A., Nielsen, C. J., Swietlicki, E., Putaud, J. P., Balkanski, Y., Fuzzi, S., Horth, J., Moortgat, G. K., Winterhalter, R., Myhre, C. E. L., Tsigaridis, K., Vignati, E., Stephanou, E. G., and Wilson, J.: Organic aerosols and global climate modeling: a review, *Atmos. Chem. Phys.*, 5, 1053–1123, 2005, <http://www.atmos-chem-phys.net/5/1053/2005/>.
- Kawamura, K., Kasukabe, H., and Barrie, L. A.: Source and reaction pathways of dicarboxylic acids, ketoacids and dicarbonyls in arctic aerosols: One year of observations, *Atmos. Environ.*, 30, 1709–1722, 1996.
- Kawamura, K., Steinberg S., and Kaplan I. R.: Homologous series of C-1-C-10 monocarboxylic acids and C-1-C-6 carbonyls in Los Angeles air and motor vehicle exhausts, *Atmos. Environ.*, 34, 4175–4191, 2000.
- Kean, A. J., Grosjean, E., Grosjean, D., and Harley, R. A.: On-Road Measurement of Carbonyls in California Light-Duty Vehicle Emissions, *Environ. Sci. Technol.*, 35, 4198–4204, 2001.
- Kroll, J. H., Ng, N. L., Murphy, S. M., Varutbangkul, V., Flagan, R. C., and Seinfeld, J. H.: Chamber studies of secondary organic aerosol growth by reactive uptake of simple carbonyl compounds, *J. Geophys. Res.*, 110, D23207, doi:10.1029/2005JD006004, 2005.
- Landgraf, J. and Crutzen, P. J.: An efficient method for online calculation of photolysis and heating rates, *J. Atmos. Sci.*, 55, 863–878, 1998.
- Lelieveld, J., Peters, W., Dentener, F. J., and Krol, M. C.: Stability of tropospheric hydroxyl chemistry, *J. Geophys. Res.*, 107, 4715, doi:10.1029/2002JD002272, 2002.
- Lee, Y. N., Zhou, X. L., and Hallock, K.: Atmospheric carbonyl compounds at a rural southeastern United States site, *J. Geophys. Res.-Atmos.*, 100, 25 933–25 944, 1995.
- Liggio, J., and McLaren, R.: An optimized method for the determination of volatile and semi-volatile aldehydes and ketones in ambient particulate matter, *Int. J. Environ. An. Ch.*, 83,

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819–835, 2003.

Liggio, J., Li, S. M., and McLaren, R.: Heterogeneous reactions of glyoxal on particulate matter identification of acetals and sulfate esters, *Environ. Sci. Technol.*, 39, 1532–1541, 2005a.

Liggio, J., Li, S. M., and McLaren, R.: Reactive uptake of glyoxal by particulate matter, *J. Geophys. Res.*, 110, D10304, doi:10.1029/2004JD005113, 2005b.

Lim, H. J., Carlton A. G., and Turpin, B. J.: Isoprene forms secondary organic aerosol through cloud processing: Model simulations, *Environ. Sci. Technol.*, 39, 4441–4446, 2005.

Loeffler, K. W., Koehler, C. A., Paul, N. M., and de Haan, D. O.: Oligomer Formation in Evaporating Aqueous Glyoxal and Methyl Glyoxal Solutions, *Environ. Sci. Technol.*, 40, 6318–6323, 2006

Matsunaga, S., Mochida, M., and Kawamura, K.: Growth of organic aerosols by biogenic semi-volatile carbonyls in the forestal atmosphere, *Atmos. Environ.*, 37, 2045–2050, 2003.

Moortgat, G. K., Grossmann, D., Boddenberg, A., Dallmann, G., Ligon, A. P., Turner, W. V., Gab, S., Slemr, F., Wieprecht, W., Acher, K., Kilber, M., Schlomski, S., Bachmann, K.: Hydrogen Peroxide, Organic Peroxides and Higher Carbonyl Compounds Determined during the BERLIOZ Campaign, *J. Atmos. Chem.* 42, 443–463, 2002.

Munger, J. W., Jacob, D. J., Daube, B. C., Horowitz, L. W., Keene, W. C., and Heikes B. G.: Formaldehyde, glyoxal, and methylglyoxal in air and cloudwater at a rural mountain site in central Virginia, *J. Geophys. Res.*, 100, 9325–9334, 1995.

Olivier, J., Peters, J., Granier, C., Petron, G., Muller, J. F., and Wallens, S.: Present and Future surface emissions of anthropogenic compounds, POET report #2, EU project EVK2-1999-00011, 2003.

Palmer, P. I., Jacob, D. J., Fiore, A. M., Martin, R. V., Chance, K., and Kurosui, T. P.: Mapping isoprene emissions over North America using formaldehyde column observations from space, *J. Geophys. Res.*, 108(D6), 4180, doi:10.1029/2002JD002153, 2003.

Poisson, N., Kanakidou, M., and Crutzen, P. J.: Impact of Non Methane Hydrocarbons on tropospheric chemistry and particular the oxidizing power of the global troposphere: 3-Dimensional Modelling results, *J. Atmos. Chem.*, 36, 157–230, 2000.

Poisson, N., Kanakidou, M., Bonsang, B., Behmann, T., Burrows, J., Gölz, C., Harder, H., Lewis, A., Moortgat, G. K., Nunes, T., Pio, C., Platt, U., Sauer, F., Schuster, G., Seakins, P., Senzig, J., Seuwen, R., Trapp, D., Voltz-Thomas, A., Zenker, T., Zitzelberger, R.: The Impact of Natural NonMethane Hydrocarbon Oxidation on the Free Radical and Ozone budgets above a Eucalyptus forest, *Chemosphere, Global Change Science*, 3, 353–366, 2001.

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- Pope, C. A., and Dockery, D.W.: Health effects of fine particulate air pollution: Lines that connect, *J. Air Waste Manage.*, 54, 709–742, 2006.
- Ramanathan, V. and Crutzen, P. J.: New directions: Atmospheric brown “Clouds”, *Atmos. Environ.*, 37, 4033–4035, 2003.
- 5 Sander, S.P., Ravishankara, A. R., Golden, D. M., Kolb, C. E., Kurylo, M. J., Molina, M. J., Moortgat, G. K., Finlayson-Pitts, B. J., Wine, P. H., Huie, R. E.: Chemical Kinetics and Photochemical Data for Use in Atmospheric Studies Evaluation Number 15, JPL Publication 06-2 (2006), available on the web at <http://jpldataeval.jpl.nasa.gov/>, 2006.
- Saxena, P. and Hildemann, L. M.: Water-Soluble Organics in Atmospheric Particles: A Critical  
 10 Review of the Literature and Application of Thermodynamics to Identify Candidate Compounds, *J. Atmos. Chem.*, 24, 57–109, 1996.
- Saunders, S. M., Jenkin, M. E., Derwent, R. G., and Pilling, M. J.: Protocol for the development of the Master Chemical Mechanism, MCM v3 (Part A): tropospheric degradation of non-aromatic volatile organic compounds, *Atmos. Chem. Phys.*, 3, 161–180, 2003
- 15 Singh, H. B., Salas, L. J., Chatfield, R. B., Czech, E., Fried, A., Walega, J., Evans, M. J., Field, B. D., Jacob, D. J., Blake, D., Heikes, B., Talbot, R., Sachse, G., Crawford, J. H., Avery, M. A., Sandholm, S., and Fuelberg, H.: Analysis of the atmospheric distribution, sources, and sinks of oxygenated volatile organic chemicals based on measurements over the Pacific during TRACE-P, *J. Geophys. Res.*, 109, D15S07, doi:10.1029/2003JD003883, 2004.
- 20 Sinreich, R.: MAX-DOAS detection of glyoxal during ICARTT 2004, *Atmos. Chem. Phys.*, 7, 1293–1303, 2007,  
<http://www.atmos-chem-phys.net/7/1293/2007/>.
- Shindell, D. T., Faluvegi, G., Stevenson, D. S., Emmons, L. K., Lamarque, J. F., Petron G., Dentener, F. J., Ellingsen K., Eskes, H., and van Noije, T. P. C.: Multi-model simulations  
 25 of carbon monoxide: Comparison with observations and projected near-future changes, *J. Geophys. Res.*, 111, D19306, doi:10.1029/2006JD007100, 2006.
- Sorooshian, A., Lu, M. L., Brechtel, F. J., Jonsson, H., Feingold, G., Flagan, R. C., and Seinfeld, J. H.: On the source of organic acid aerosol layers above clouds, *Environ. Sci. Technol.*, 41, 4647–4654, 2007.
- 30 Spaulding, R. S., Schade, G. W., Goldstein, A. H., and Charles, M. J.: Characterization of secondary atmospheric photooxidation products: Evidents for biogenic and anthropogenic sources, *J. Geophys. Res.*, 108(D8), 4247, doi:10.1029/2002JD002478, 2003.
- Stevenson, D. S., Dentener, F. J., Schultz, M. G., Ellingsen, K., and van Noije, T. P. C.: Multi-

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- model ensemble simulations of present-day and near-future tropospheric ozone. *J. Geophys. Res.*, 111, D08301, doi:10.1029/2005JD006338, 2006.
- Tadić, J., Moortgat, G. K., and Wirtz, K.: Photolysis of glyoxal in air, *J. Photoch. Photobiol. A*, 177, 116–124, 2006.
- 5 Tsigaridis, K. and Kanakidou, M.: Global modeling of secondary organic aerosol in the troposphere: a sensitivity analysis, *Atmos. Chem. Phys.* 3, 1849–1869, 2003.
- Tsigaridis, K., Krol, M., Dentener, F. J., Balkanski Y., Lathiere J., Metzger S., Hauglustaine, D. A., and Kanakidou, M.: Change in global aerosol composition since preindustrial times, *Atmos. Chem. Phys.*, 6, 5143–5162, 2006,
- 10 <http://www.atmos-chem-phys.net/6/5143/2006/>.
- Tsigaridis, K. and Kanakidou, M.: Secondary organic aerosol importance in the future atmosphere, *Atmos. Environ.* doi:10.1016/j.atmosenv.2007.03.045, 2007.
- van Noije, T. P. C., van Eskes, H. J., van Weele, M., and van Velthoven, P. F. J.: Implications of the enhanced Brewer-Dobson circulation in European Centre for Medium-Range Weather
- 15 Forecasts reanalysis ERA-40 for the stratosphere-troposphere exchange of ozone in global chemistry transport models, *J. Geophys. Res.*, 109, D19308, doi:10.1029/2004JD004586, 2004.
- van Noije, T. P. C. van Eskes H. J., Dentener, F. J., Stevenson, D. S., Ellingsen, K., Schultz, M. G., and Wild, O.: Multi-model ensemble simulations of tropospheric NO<sub>2</sub> compared with GOME retrievals for the year 2000, *Atmos. Chem. Phys.*, 6, 2943–2979, 2006,
- 20 <http://www.atmos-chem-phys.net/6/2943/2006/>.
- Velasco, E., Lamb, B., Westberg, H., Allwine, E., Sosa, G., Arriaga-Colina, J. L., Jobson, B. T., Alexander, M. L., Prazeller, P., Knighton, W. B., Rogers, T. M., Grutter, M. Herndon, S. C., Kolb, C. E., Zavala, M., de Foy, B., Volkamer, R., Molina, L. T., and Molina, M. J.: Distribution,
- 25 magnitudes, reactivities, ratios and diurnal patterns of volatile organic compounds in the Valley of Mexico during the MCMA 2002 & 2003 field campaigns, *Atmos. Chem. Phys.*, 7, 329–353, 2007,
- <http://www.atmos-chem-phys.net/7/329/2007/>.
- Volkamer, R., Platt, U., and Wirtz, K.: Primary and secondary glyoxal formation from aromatics: experimental evidence for the bicycloalkyl-radical pathway from benzene, toluene, and p-xylene, *J. Phys. Chem. A*, 105, 7865–7874, 2001.
- 30 Volkamer, R., Barnes, I., Platt, U., Molina, L. T., and Molina, M. J.: Remote Sensing of Glyoxal by Differential Optical Absorption Spectroscopy (DOAS): Advancements in simulations

## Secondary sources and distribution of glyoxal

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chambers and field experiments, paper presented at Advanced Research Workshop on Environmental Simulation Chambers: Application to Atmospheric Chemical Processes, NATO, Zakopane, Poland, 1–4 October, 2005a.

Volkamer, R., Molina, L. T., Molina, M. J., Shirley, T., and Brune, W. H.: DOAS measurement of glyoxal as an indicator for fast VOC chemistry in urban air, *Geophys. Res. Lett.*, 32, L08806, doi:10.1029/2005GL022616, 2005b.

Volkamer, R., Spietz, P., Burrows, J., Platt U.: High-resolution absorption cross-section of glyoxal in the UV–vis and IR spectral ranges, *J. Photoch. Photobiol. A*, 172, 35–46, 2005c.

Volkamer, R., Jimenez, J. L., Martini, F. S., Dzepina, K., Zhang, Q., Salcedo, D., Molina, L. T., Worsnop, D. R., and Molina, M. J.: Secondary organic aerosol formation from anthropogenic air pollution: Rapid and higher than expected, *Geophys. Res. Lett.*, 33, L17811, doi:10.1029/2006GL026899, 2006a.

Volkamer, R., Kurosu, T. P., Chance, K., Li, Z., Zhang, Y., Brauers, T., and Wahner, A.: Spatial Variability of Glyoxal, HCHO and NO<sub>2</sub> during PRD-2006: Comparison of mobile Mini-MAX-DOAS and OMI satellite data in the Pearl River Delta, China. *Eos Trans. AGU*, 87(52), Fall Meet. Suppl., Abstract A31B-0897, 2006b.

Volkamer, R., San Martini, F., Salcedo, D., Molina, L. T., Jimenez, J. L., and Molina, M. J.: A Missing Sink for Gas-Phase Glyoxal in Mexico City: Formation of Secondary Organic Aerosol, *Geophys. Res. Lett.*, 34, L19807, doi:10.1029/2007GL030752, 2007a.

Vountas, M., Dinter, T., Bracher, A., Burrows, J. P., and Sierk, B.: Spectral studies of ocean water with space-borne sensor SCIAMACHY using Differential Optical Absorption Spectroscopy (DOAS), *Ocean Sci.*, 3, 429–440, 2007, <http://www.ocean-sci.net/3/429/2007/>.

Vrekoussis, M., Kanakidou, M., Mihalopoulos, M., Crutzen, P. J., Lelieveld, J., Perner, D., Berresheim, H., and Baboukas, E.: Role of the NO<sub>3</sub> radicals in oxidation processes in the eastern Mediterranean troposphere during the MINOS campaign, *Atmos. Chem. Phys.*, 4, 169–182, 2004, <http://www.atmos-chem-phys.net/4/169/2004/>.

Wittrock, F., Oetjen, H., Richter, A., Fietkau, S., Medeke, T., Rozanov, A., and Burrows, J. P.: MAX-DOAS measurements of atmospheric trace gases in Ny-Alesund-Radiative transfer studies and their application, *Atmos. Chem. Phys.*, 4, 955–966, 2004, <http://www.atmos-chem-phys.net/4/955/2004/>.

Wittrock, F., Richter, A., Oetjen, H., Burrows, J.P., Kanakidou, M., Myriokefalitakis, S., Volkamer,

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R., Beirle, S., Platt, U., and Wagner, T.: Simultaneous Global Observations of Glyoxal and Formaldehyde from Space, Geophys. Res. Lett., 33, L16804, doi:10.1029/2006GL026310, 2006.

- 5 Wesely, M. L.: Parameterization of surface resistances to gaseous dry deposition in regional scale numerical models, Atmos. Environ., 23, 1293–1304, 1989.

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## Secondary sources and distribution of glyoxal

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**Table 1.** Reactions taken into account in the TM4 model for CHOCHO production and destruction and references for the rate constants adopted in the model.

a)			
Num	Reactions (Gas phase chemical production)	k	Reference for k
1	Isoprene + OH → 0.03CHOCHO + products	2.7E-11exp(390/T)	IUPAC (2006)
2	C <sub>2</sub> H <sub>4</sub> + O <sub>3</sub> + 2O <sub>2</sub> → 0.0044CHOCHO + products	3.3E-12exp(-2880/T)	IUPAC (2006)
3	C <sub>3</sub> H <sub>6</sub> + O <sub>3</sub> + 2O <sub>2</sub> → 0.05CHOCHO + products	4.6E-13exp(-1155/T)	IUPAC (2006)
4	C <sub>2</sub> H <sub>2</sub> + OH → 0.635CHOCHO + products	5E-30(T/300) <sup>-1.5</sup> [N <sub>2</sub> ]	IUPAC (2006)
5	Benzene + OH → 0.36CHOCHO + products	2.47E-12exp(207.0/T)	Calvert et al. (2002)
6	Toluene + OH → 0.306CHOCHO + products	5.96E-12	Atkinson (1994)
7	Xylene + OH → 0.319CHOCHO + products	1.72E-11	Average of ortho-, meta-, para-isomers
8	HOCH <sub>2</sub> CHO + OH + O <sub>2</sub> → CHOCHO + HO <sub>2</sub> + H <sub>2</sub> O	2.2E-12	IUPAC (2006)
b)			
Num.	Reactions (Gas phase chemical destruction)	k	Reff
1	CHOCHO + OH → 2CO + HO <sub>2</sub> + H <sub>2</sub> O	1.44E-12exp(-1862/T)	Feierabend et al. (2008)
2	CHOCHO + NO <sub>3</sub> + O <sub>2</sub> → HNO <sub>3</sub> + HO <sub>2</sub> + 2CO	1.2E-15	upper limit estimate by I. Barnes (personal communication, 2007).
3	CHOCHO + hν → 2HO <sub>2</sub> + 2CO	J <sub>CHOCHO</sub>	IUPAC (2006)

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**Table 2.** Glyoxal TM4 model calculations evaluation against ground based observations.

Location	Coordinates	Period	Model <i>ppt: part per trillion (0.001 ppb).</i>	Observations	Reference
Pinnacles, Shenandon National Park	36.29 N 121.10 W	Autumn	0.2 ppb	0.04 ppb	Munger et al. (1995)
Georgia, USA	32.32 N 82.08 E	Summer	0.2 ppb	0.02–0.09 ppb	Lee et al. (1995)
Long Beach, Central LA, Azusa, Claremont, USA	33.4 N 118.11 W (mean)	Autumn	0.16 ppb	0.78±0.85 ppb	Grosjean et al. (1996)
Porto Allegre, Brazil	30.2 S 51.1 W	Spring	0.03 ppb	0.3 ppb	Grosjean et al. (1999)
Los Angeles, USA	34.3 N 118.15 W	Autumn	0.16 ppb	0.04–0.95 ppb	Kawamura et al. (2000)
Giesta, Portugal	41.11 N 8.35 E	Summer	0.11 ppb	0.52–2.42 ppb	Borrego et al. (2000)
Las Vegas, USA	36.10 N 115.12 W	Summer	0.12 ppb	0.12–0.42 ppb	Jing et al. (2001)
Hong Kong, China	22.18 N 114.1 E	Winter	0.03 ppb	0.09–0.21 ppb	Ho and Yu (2002)
		Winter	0.06 ppb	19.7–28.0 ppb	
				8.4–23.7 ppb	
				8.6–11.9 ppb	
Pabsthum (near Berlin), Germany	52.51 N 12.56 E	Summer	0.2 ppb	8.8–14.8 ppb	Moortgat et al. (2002)
Mexico City, Mexico	19.24 N 99.12 W	Spring	0.2 ppb	0.01–0.12 ppb	
MIT, USA	42.36 N 71.09 W	Summer	363 ppt	0.15–1.82 ppb	Volkamer et al. (2005)
ICARTT 2004	43 N	Summer	141 ppt	40–140 ppt	Sinreich et al. (2007)
RV Ron Brown	68 W			100–350 ppt	Sinreich et al. (2007)

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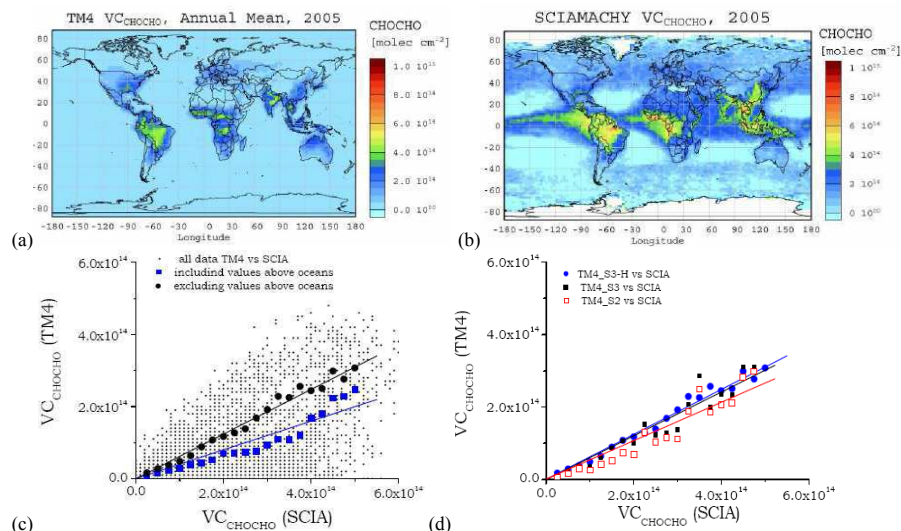
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**Fig. 1.** Global annual mean column distribution of CHOCHO (in 2°×3° grid) for the year 2005 (in molecules cm<sup>-2</sup>):

- (a)** Simulated by TM4 by taking into account all glyoxal sources (simulation S3-H).
- (b)** Observed by the satellite based sensor SCIAMACHY
- (c)** Comparison of annual mean glyoxal columns from TM4- S3-H simulations and SCIAMACHY observations (in molecules cm<sup>-2</sup>). Points show annual mean data in 2°×3° grid for the year 2005. Solid blue squares are the corresponding binned data (see text). Solid black circles are binned data when neglecting the grids over the oceans.
- (d)** Comparison of annual mean glyoxal columns from TM4 simulations and SCIAMACHY observations (in molecules cm<sup>-2</sup>). Binned data over the continents; solid blue circles are for S3-H simulations, solid black squares for S3 simulation and open red squares for S2 simulation. Linear fit regression lines are also plotted with the corresponding color.

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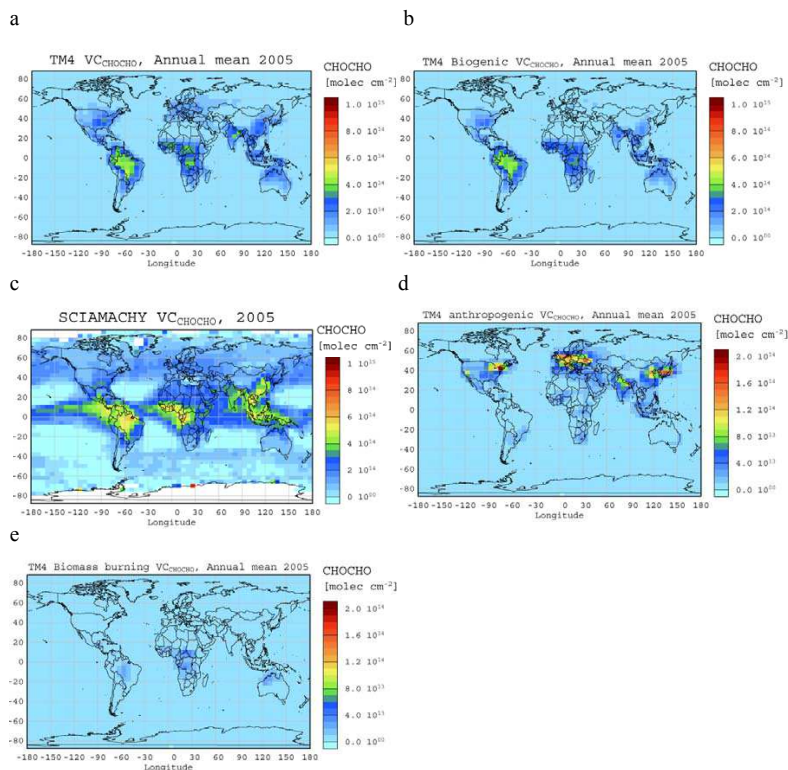
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**Fig. 2.** Global annual mean column distribution of CHOCHO (in 4°x6° grid) for the year 2005 (in molecules cm<sup>-2</sup>):

- (a) Simulated by TM4 by taking into account all glyoxal sources (simulation S3).
- (b) Simulated by TM4 to be produced by the biogenic VOC oxidation.
- (c) Observed by the satellite based sensor SCIAMACHY (reduced spatial resolution)
- (d) Simulated by TM4 to be produced by the anthropogenic VOC oxidation
- (e) Simulated by TM4 to be produced by the biomass burning VOC oxidation

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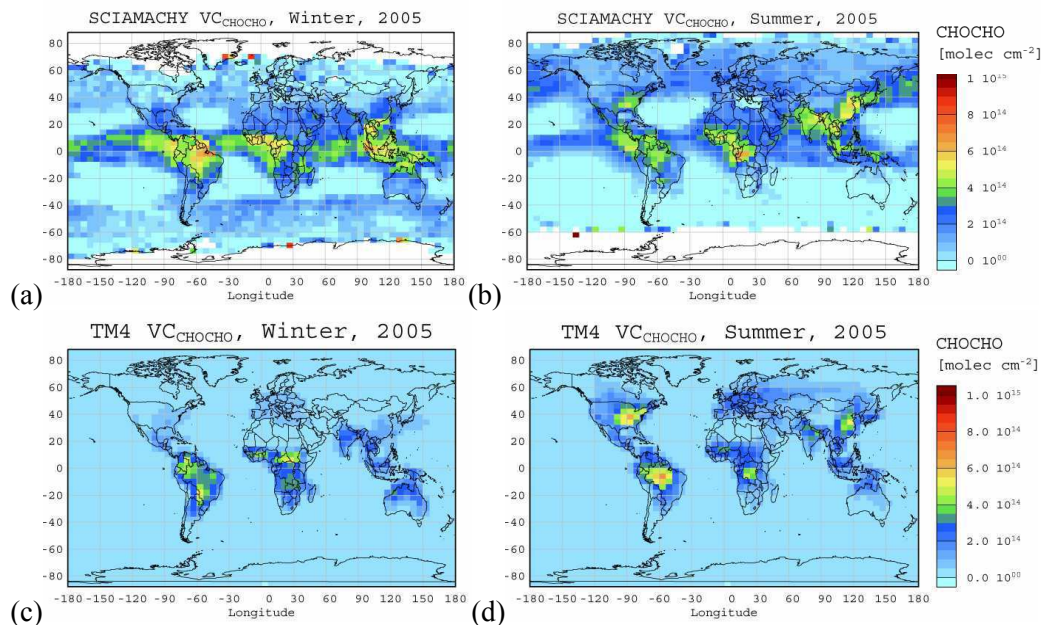
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**Fig. 3.** NH winter (a, c, e, g, i) and NH summer (b, d, f, h, j) mean columns of glyoxal as derived from SCIAMACHY observations (a, b) and as computed by TM4 taking into account formation of glyoxal from VOC emitted from (c, d) all sources; (e, f) only biogenic sources; (g, h) only anthropogenic sources and (e, f) only biomass burning for the year 2005. Columns are given in  $10^{15}$  molecules  $\text{cm}^{-2}$ . Note different color scale in figures (g) to (j).

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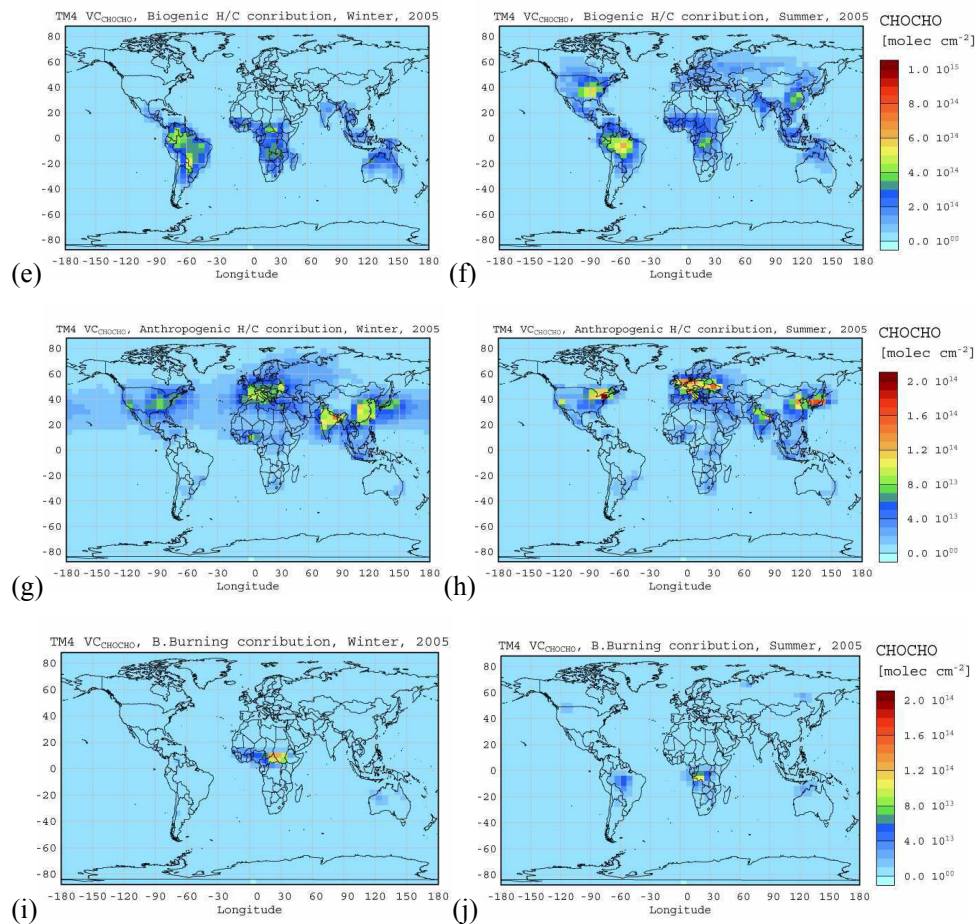
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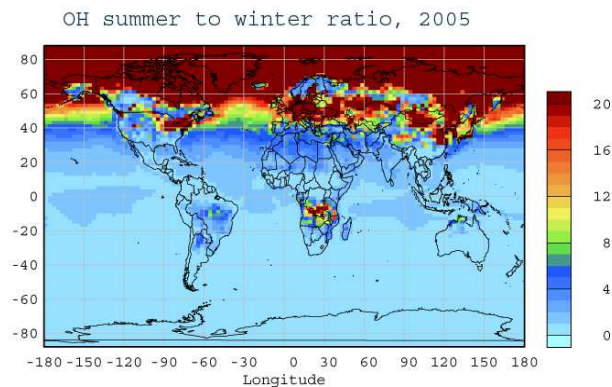
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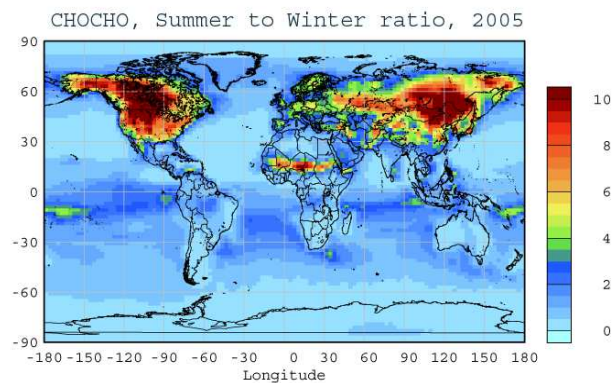
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**Fig. 3.** Continued.

(a)



(b)



**Fig. 4.** NH Summer to NH winter ratios of **(a)** computed OH radical, **(b)** of computed glyoxal column (simulation S3-H).

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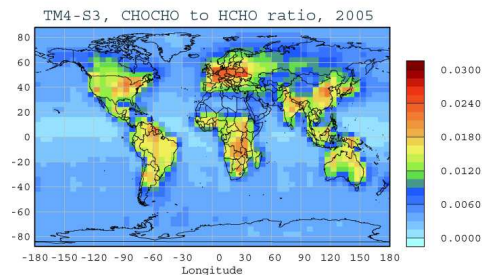
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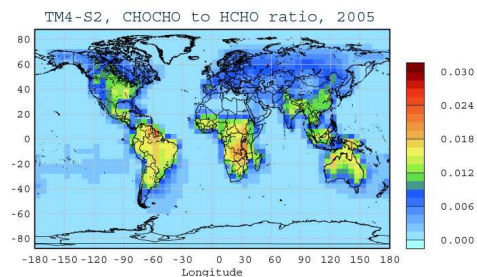
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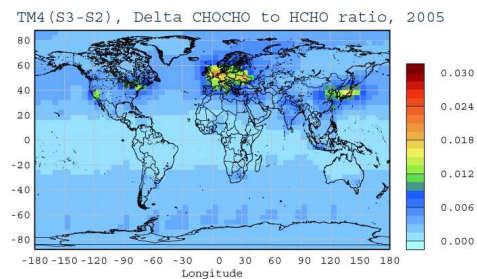
(a)



(b)



(c)



**Fig. 5.** Annual mean ratios of CHOCHO to HCHO as simulated by TM4 **(a)**: simulation S3 and **(b)**: simulation S2. **(c)** Difference in this ratio attributed to the anthropogenic contribution to the CHOCHO columns.

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